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Class Date Problem Solving continued Sample Problem 1 Ammonia is  
made industrially by reacting nitrogen and hydrogen under pressure, at  
high temperature, and in the presence of a catalyst. The equation is  $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$ . If 4.0 mol of  $H_2$  react, how many moles of  $NH_3$  will be produced?

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Class Date Problem Solving continued CONVERTING NUMBER  
OF ATOMS OF AN ELEMENT TO MASS Sample Problem 4 uses  
the progression of steps 1 2 3 to convert from the mass of an  
element to the number of atoms. In order to calculate the mass from a  
given number of atoms, these steps will be reversed.



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stoichiometry so far in your chemistry course you have learned that chemists count quantities of elements and compounds in terms of moles and that they relate moles of a substance to mass by using the molar mass in addition you have learned to write chemical equations so that they represent the rearrangements of atoms that take place

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With such a wide diversity of properties and applications, is it any wonder that industry and academia have such a fascination with polymers? A solid introduction to such an enormous and important field is critical to the modern polymer scientist-to-be, but most of the available books do not stress practical problem solving or include recent advances. Serving as the polymer book for the new millennium, *Introduction to Polymer Science and Chemistry: A Problem Solving Approach* unites the fundamentals of polymer science and polymer chemistry in a seamless presentation. Emphasizing polymerization kinetics, the author uses a unique question-and-answer approach when developing theory or introducing new concepts. The first four chapters introduce polymer science, focusing on physical and



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molecular properties, solution behavior, and molecular weights. The remainder of the book explores polymer chemistry, devoting individual, self-contained chapters to the main types of polymerization reactions: condensation; free radical; ionic; coordination; and ring-opening. It introduces recent advances such as supramolecular polymerization, hyperbranching, photoemulsion polymerization, the grafting-from polymerization process, polymer brushes, living/controlled radical polymerization, and immobilized metallocene catalysts. With numerical problems accompanying the discussion at every step along with numerous end-of-chapter exercises, *Introduction to Chemical Polymer Science: A Problem Solving Approach* is an ideal introductory text and self-study vehicle for mastering the principles and methodologies of modern polymer science and chemistry.

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