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chlorofluorocarbons, CFC ' s 1.2

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Chapter Summaries – Chemistry
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Chapter 1 Section 2 MATTER and Its
Properties Matter – Anything that
has MASS and VOLUME. ATOMS and
MOLECULES the basic building blocks
of matter. ATOM-- the smallest unit of
an element that still maintains the
properties that element. ELEMENT–
a pure substance made up of only one
type of atom.

MATTER and CHANGE Chapter 1
Section 1

Chemistry: Matter and Change
Chapter 6 Study Guide for Content
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MASTERY Section 6.3 Periodic Trends

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In your textbook, read about atomic radius and ionic radius. Circle the letter of the choice that best completes the statement or answers the question. 1. Atomic radii cannot be measured ...

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Section 3.1 Properties of Matter pages
70–75 Problem-Solving Lab 1.

Explain why the flow of a compressed
gas must be controlled for practical
and safe use.

Matter—Properties and

Changes Matter—Properties and

Changes

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Table and Periodic Law Section 6.1
Development of the Modern Periodic
Table 1. octaves 2. eight 3. nine 4.
accepted 5. Dmitri Mendeleev 6.
atomic mass 7. elements 8. atomic
number 9. Henry Moseley 10. protons
11. periodic law 12. properties 13. 14
...

Ch 6 Study Guide answers
Date Class Chapter Assessment
Chemistry: Matter and Change •
Chapter 5 25 Electrons in Atoms
Reviewing Vocabulary Match the
definition in Column A with the term
in Column B. Column A Column B 1.
The set of frequencies of the
electromagnetic waves emitted by
the atoms of an element 2. The

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minimum amount of energy that can
be lost or gained by ...

Chapter Assessment

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q... the study of energy changes (that
is, energy produced or absor... A
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Change. extensive property. intensive
property. extensive properties
(examples) intensive property
(examples) a property that depends
on the amount of matter in a sample.
a property that depends on the type
of matter in a sample.... mass,
volume, length, weight.

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Chapter 11 31 number of moles of
known substance C H C H C H O O O
CO CO CO H O H O H O number of
moles of unknown substance O CO H
O C H CO H O C H O H O C H O CO use
mole ratio 5 mol O 2 1 mol C 3 H 8 3
mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1
mol C 3 H 8 1 mol C 3 H 8 5 mol O 2

MATH SKILLS TRANSPARENCY MASTER 16

A “ chemical change ” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change. Properties of Compounds Quite different properties than their component elements.

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